

## Niosh Pocket Guide Chlorine

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Greenish-yellow gas with a pungent, irritating odor. [Note: Shipped as a liquefied compressed gas.] Molecular Weight.

**CDC—NIOSH Pocket Guide to Chemical Hazards—Chlorine**

Physical Description Yellow to red gas or a red-brown liquid (below 52 ° F) with an unpleasant odor similar to chlorine and nitric acid.

**CDC—NIOSH Pocket Guide to Chemical Hazards—Chlorine—**

Physical Description. Colorless gas or a greenish-yellow liquid (below 53 ° F) with a somewhat sweet, suffocating odor. [Note: Shipped as a liquefied compressed gas.] Nonflammable Gas Noncombustible Liquid, but contact with organic materials may result in SPONTANEOUS ignition.

**CDC—NIOSH Pocket Guide to Chemical Hazards—Chlorine—**

Colorless to pale-yellow, viscous liquid or solid (below 50 ° F) with a mild, hydrocarbon odor. Molecular Weight. 326 (approx) Boiling Point. 689-734 ° F. Freezing Point. 50 ° F. Solubility.

**CDC—NIOSH Pocket Guide to Chemical Hazards—**

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Bookmark File PDF Niosh Pocket Guide Chlorine (hydrate, frozen) Formula ClO 2: Conversion 1 ppm = 2.76 mg/m 3: IDLH 5 ppm See: 10049044: Exposure Limits NIOSH REL Niosh Pocket Guide Chlorine NIOSH. Up to 5 ppm: (APF = 10) Any chemical cartridge respirator with cartridge(s) providing protection against the compound of concern\* (APF = 10)

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niosh At concentrations above the NIOSH REL, or where there is no REL, at any detectable concentration: (APF = 10,000) Any self-contained breathing apparatus that has a full facepiece and is operated in a pressure-demand or other positive-pressure mode

**CDC—NIOSH Pocket Guide to Chemical Hazards—Chlordane**

The NIOSH Pocket Guide to Chemical Hazards (NPG) informs workers, employers, and occupational health professionals about workplace chemicals and their hazards. The NPG gives general industrial hygiene information for hundreds of chemicals/classes. The NPG clearly presents key data for chemicals or substance groupings (such as cyanides, fluorides, manganese compounds) that are found in workplaces.

**Pocket Guide to Chemical Hazards—NIOSH—CDC**

What chlorine is . Chlorine is an element used in industry and found in some household products. ... National Institute for Occupational Safety and Health (NIOSH), Pocket Guide to Chemical Hazards; Page last reviewed: April 4, 2018. Content source: National Center for Environmental Health (NCEH) Specific Hazards. Bioterrorism. A-Z. Anthrax ...

**CDC—Facts About Chlorine**

Chlorine Synonyms & Trade Names Molecular chlorine CAS No. 7782-50-5 RTECS No. FO2100000: DOT ID & Guide 1017 124: Formula Cl 2: Conversion 1 ppm = 2.90 mg/m 3: IDLH 10 ppm See: 7782505: Exposure Limits NIOSH REL: C 0.5 ppm (1.45 mg/m 3) [15-minute] OSHA PEL

**CDC—NIOSH Pocket Guide to Chemical Hazards—Chlorine**

Download File PDF Niosh Pocket Guide Chlorine Formula Cl 2: Conversion 1 ppm = 2.90 mg/m 3: IDLH 10 ppm See: 7782505: Exposure Limits NIOSH REL: C 0.5 ppm (1.45 mg/m 3) [15-minute] OSHA PEL CDC - NIOSH Pocket Guide to Chemical Hazards - Chlorine CDC - NIOSH Pocket Guide to Chemical Hazards - Chlorodiphenyl (54% chlorine)

**Niosh Pocket Guide Chlorine—rmapl.youthmanual.com**

In case of a spill or leak, handlers must wear chemical-resistant gloves (such as nitrile or butyl) and a full-face canisterstyle (gas mask) respirator with a canister approved for chlorine (MSHA/NIOSH approval number prefix TC-14G) OR a self-contained breathing apparatus (SCBA) (MSHA/NIOSH approval number prefix TC-13F).

**Chlorine—Protective Equipment—Clothing**

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View odor for chlorine in WISER.

**Chlorine—Odor**

NIOSH 6011 Chlorine— Products and Materials Guide | Sigma ... The NIOSH Pocket Guide to Chemical Hazards (NPG) informs workers, employers, and occupational health professionals about workplace chemicals and their hazards. The NPG gives general industrial hygiene information for hundreds of chemicals/classes.

**Niosh Pocket Guide Chlorine—veksite.cz**

Chlorine persists as an element only at a very low pH (less than 2), and at the higher pH found in living tissue it is rapidly converted into hypochlorous acid. In this form, it apparently can penetrate the cell and form N-chloro-derivatives that can damage cellular integrity.

**Chlorine—Other Properties**

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