

Intermolecular Forces Pogil Answers

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Intramolecular forces occur within/inside molecules, while intermolecular forces occur between molecules.

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There are three different types of intermolecular forces: dipole-dipole interactions hydrogen bonding (although technically this is not a bond because it does not involve electrons being shared or transferred), and dispersion forces..

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Model 2: What are the three types of intermolecular forces? There are three different types of intermolecular forces: dipole-dipole interactions, hydrogen bonding (although technically this is not a bond because it does not involve electrons being shared or transferred), and dispersion forces. Figure 2: Dipole-Dipole Interactions

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We will consider the following types of intermolecular forces: London dispersion, dipole-dipole, and hydrogen bonding. London dispersion forces and dipole-dipole forces are collectively known as van der Waals forces. Molecules can have any mix of these three kinds of intermolecular forces, but all substances at least have London dispersion forces.

[Chem 116 POGIL Worksheet - Week 3 Intermolecular Forces](#)

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POGIL: Intermolecular Forces and Boiling Points Model 1: Intermolecular Forces in Liquids and Gases Molecules attract each other, and the intermolecular force increases rapidly as the distance between the molecules decreases. In a liquid, the molecules are very close to one another and are constantly moving and colliding.

[POGIL: Intermolecular Forces and Boiling Points](#)

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Intermolecular Forces Exercises. Answer the following to the best of your ability. Questions left blank are not counted against you. When you have completed every question that you desire, click the "MARK TEST" button after the last exercise. A new page will appear showing your correct and incorrect responses.

[Intermolecular Forces Exercises](#)

The Intermolecular Forces (forces between molecules) are weaker than Intramolecular . Forces (The Chemical Bonds within an Individual Molecule). This distinction is the . reason we define the molecule in the first place. The properties of matter result from the properties of the individual molecule (resulting from chemical bonding) and how the molecules act collectively (resulting from intermolecular forces).

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Intermolecular Forces (A)Identify the intermolecular forces present in the following substances, and (B) select the substance with the highest boiling point: CH₃CH₃, CH₃OH, and CH₃CH₂OH Answers: (a) CH₃CH₃ has only dispersion forces, whereas the other two substances have both dispersion forces and hydrogen bonds; (b) CH₃CH₂OH