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form. It is exemplified at its simplest when steam from a kettle becomes deposited as drops of distilled water on a cold surface.

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Distillation is a process in which a liquid mixture of volatile components is separated by imparting energy to it in consideration with the boiling points of the components so that selective vaporization takes place. This process can also be used in reverse to selectively condense the

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vapour mixture.

What Is Distillation? - Chemical Engineering World

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steam from a kettle becomes deposited as drops of distilled water on a cold surface. Distillation is used to separate liquids from nonvolatile solids, as in the separation of alcoholic liquors from fermented materials, or in the separation of two or more liquids having different boiling points, as in the separation of gasoline, ...

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Z. Lei, in Reference Module in Chemistry, Molecular Sciences and Chemical Engineering, 2017. Introduction.

Distillation is the most commonly used method for the separation of homogeneous liquid mixtures and is based on differences in the boiling points or relative volatility of the constituent components. 1,2

However, where the relative volatility is

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close to unity (e.g., for the separation of azeotropic mixtures or close boiling components), a third component (i.e., entrainer, solvent, or ...

Distillation - an overview | ScienceDirect Topics

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Distillation refers to the selective boiling and subsequent condensation of a component in a liquid mixture. It is a separation technique that can be used to either increase the concentration of a particular component in the mixture or to obtain (almost) pure components from the mixture.

Distillation - Definition, Detailed Process, Types, Uses

Distillation is the process of separating the components or substances from a liquid mixture by using selective boiling and condensation. Distillation may result in essentially complete separation (nearly

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pure components), or it may be a partial separation that increases the concentration of selected components in the mixture.

Distillation - Wikipedia

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