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Chemistry Chapter 2 Test. STUDY. PLAY. Pure Substance. matter that has a definite and uniform composition. Physical Property. quality of a substance that can be observed without changing the substances composition. Extensive Properties. dependent upon the amount of substance present

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2-nitrite SO 3-2 sulfite HSO 3-bisulfite SO 4-2 sulfate HSO 4-bisulfate HCO 3-bicarbonate CO 3-2 carbonate CH 3CO 2-Acetate HPO 4 2-hydrogen phosphate H 2PO 4-dihydrogen phosphate PO 4-3 Phosphate ClO-hypochlorite ClO 2-chlorite ClO 3 - chlorate ClO 4-perchlorate MnO 4 - Permanganate CrO 4-2 Chromate Cr 2O 7-2 Dichromate You must memorize ALL ...

~~Chapter 2 Atoms, Molecules, and Ions—Chemistry~~

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~~Chapter 2: Chemistry Comes Alive~~

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difference between mass and weight. Describe the difference between chemical and physical change. Describe the difference between accuracy and precision. Describe the five states of matter, and give an example of each

~~AP Chemistry — Chapters 1 and 2~~

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Be able to discuss at least six steps of the process. 2) Be able to compare and contrast scientific law and scientific theory. Give a couple examples of each one. Section 2.1: Properties of Matter. 3) Distinguish between the physical properties and chemical properties of matter. 4) Explain the gas, liquid, and solid states of matter in terms of the kinetic molecular theory and physical properties.

~~CHAPTER 1 CHAPTER 2 — SharpSchool~~

ORGANIC CHEMISTRY I CHEM 2323 Fall 2014 Instructor: Michael C. Biewer biewerm@utdallas.edu, BE 3.326 Syllabus.

Lecture Notes: The following material can be downloaded as pdf files. Material for test #1 . Chapter 1 Chapter 2 Chapter 3 Chapter 5 . Material for test #2 . Chapter 4

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